



Matériaux pour la catalyse : Veille et prospective

Eric Marceau

CARMEN EVOLUTION – 21.06.22



“Matériaux pour la catalyse: veille”

Veille sur... des matériaux innovants?

*... des matériaux pour
des procédés innovants?*

...des synthèses innovantes?

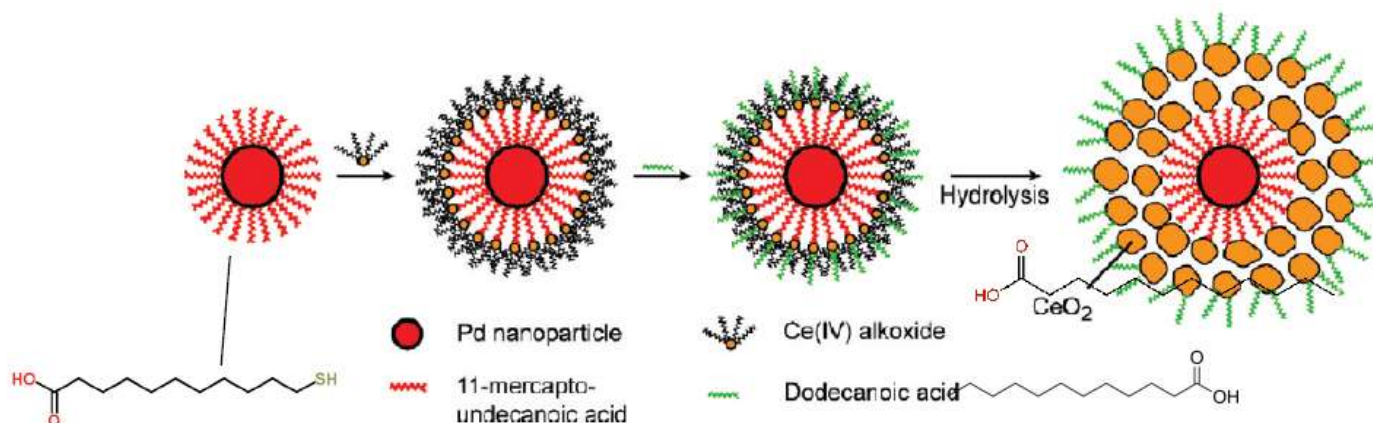
X. Carrier

CAT PREP
Summer School
2019

Synthesis of Pd@CeO₂ core-shell nanostructures

M. Cargnello, N.L. Wieder, T. Montini, R.J. Gorte, P. Fornasiero *J. Am. Chem. Soc.* 2010
M. Cargnello, J.J. Delgado Jaén, J.C. Hernandez-Garrido, K. Bakhmutsky, T. Montini, J.J. Calvino Gamez, R.J. Gorte, P. Fornasiero *Science* 2012

- 4 steps synthesis :
- (1) Synthesis of thiolate-protected Pd nanoparticles
 - (2) Self-assembly of Ce(IV) alkoxide on protected Pd NPs
 - (3) Hydrolysis of the Ce alkoxide → Pd@CeO₂ nanostructures



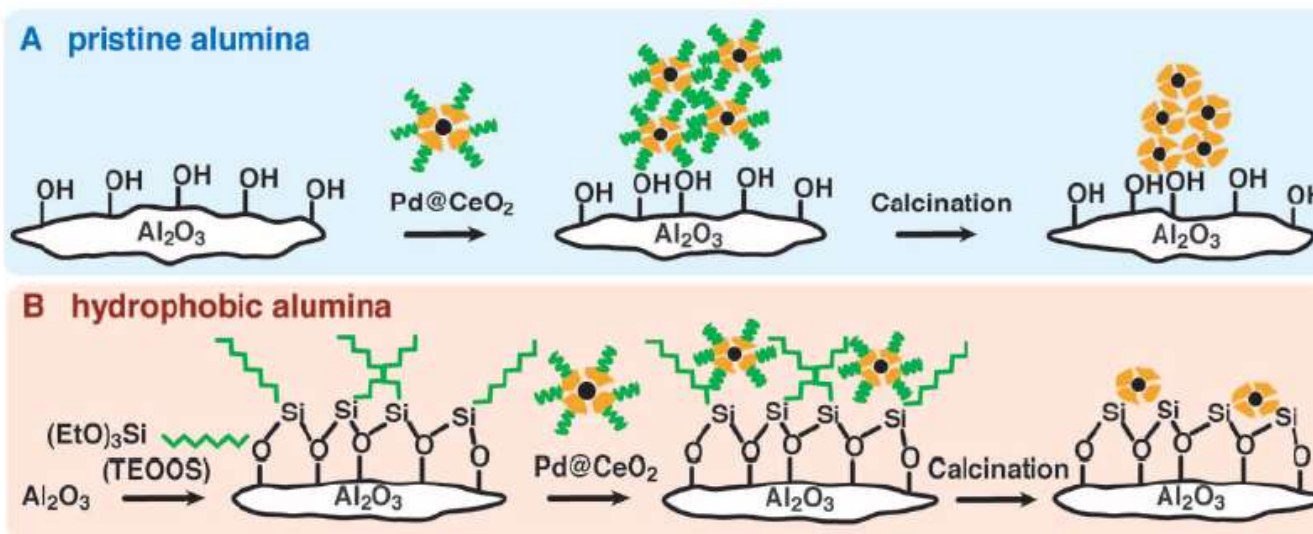
X. Carrier

CAT PREP
Summer School
2019

Deposition on alumina

Weak interaction between hydrophobic Pd@CeO₂ structures and hydrophilic alumina surface

→ Functionalisation of alumina with organosilane





“Matériaux pour la catalyse: veille”

X. Carrier

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2019

Versatile synthesis :
variation of Pd NPs size & thickness of CeO₂ shell

- (1) Synthesis: *Coût?*
- (2) Synthesis: *Rendement?*
- (3) Self-assembly: *Sécurité?*
- (4) Deposition: *Sécurité?*

nanoparticles : 5 steps

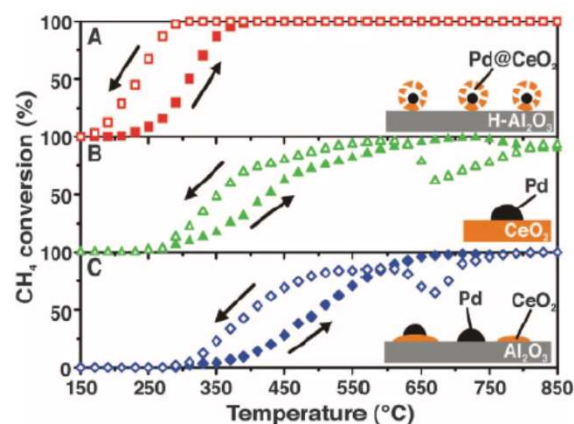
layer : 8 steps

protected Pd NPs and hydrolysis : 2 steps

more than 10 steps

Use of organic (potentially toxic) solvents : toluene, dichloromethane, methanol....

But highly active catalyst (methane combustion)



- ✓ Accessible Pd NPs (CO chemisorption)
- ✓ Complete conversion 300°C lower than other systems :
 - High resistance of Pd toward sintering due to the oxide shell
 - Enhanced CeO₂-Pd interaction improves the stabilization of the active PdO_x phase

19



“Matériaux pour la catalyse: veille”

QUELS CRITERES DE CHOIX RETENIR?

*Exhaustivité?
Virtuosité scientifique?
Applicabilité?*

Objectif ou subjectif?



“Matériaux pour la catalyse: veille”

Choix : la subjectivité, du particulier au général

Fil rouge: le projet NobleFreeCat



Mise en exergue: tout ce que
nous **N’AVONS PAS** su faire



Verrou 1

Economie de la synthèse:

Matières premières
Saut d'échelle
Recyclage



Economie de la synthèse

FCCAT 2022: “Mechanism investigation and support optimization for the **iridium**-catalyzed **hydrodeoxygenation of phenols**”

Matériau catalytique « **durable** »?



Economie de la synthèse

Métaux nobles : les disperser au maximum au moyen de synthèses supportant le changement d'échelle

ARTICLES

<https://doi.org/10.1038/s41565-021-01022-y>

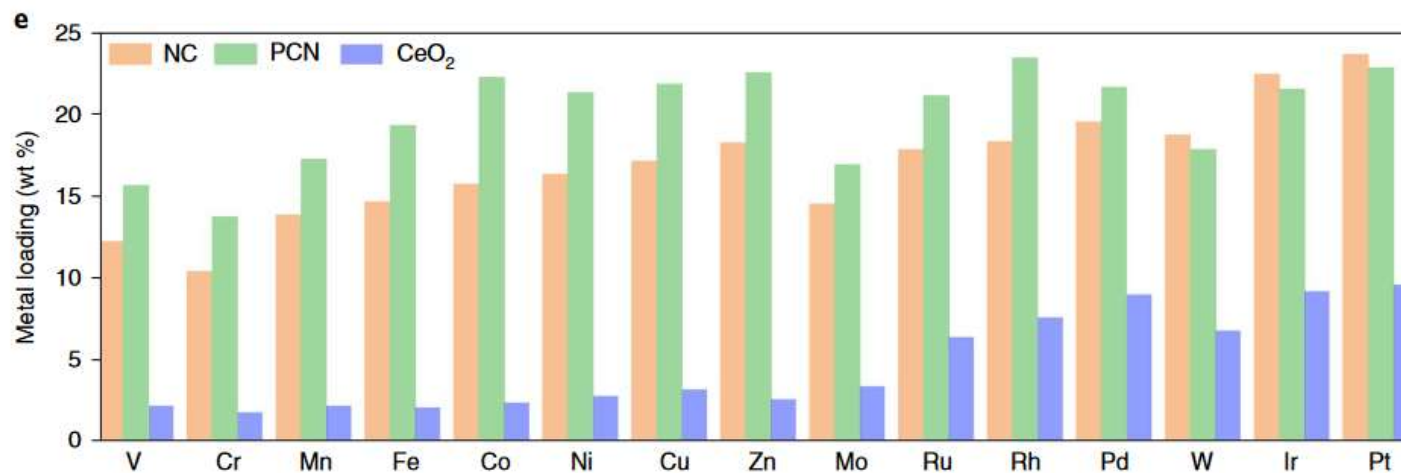
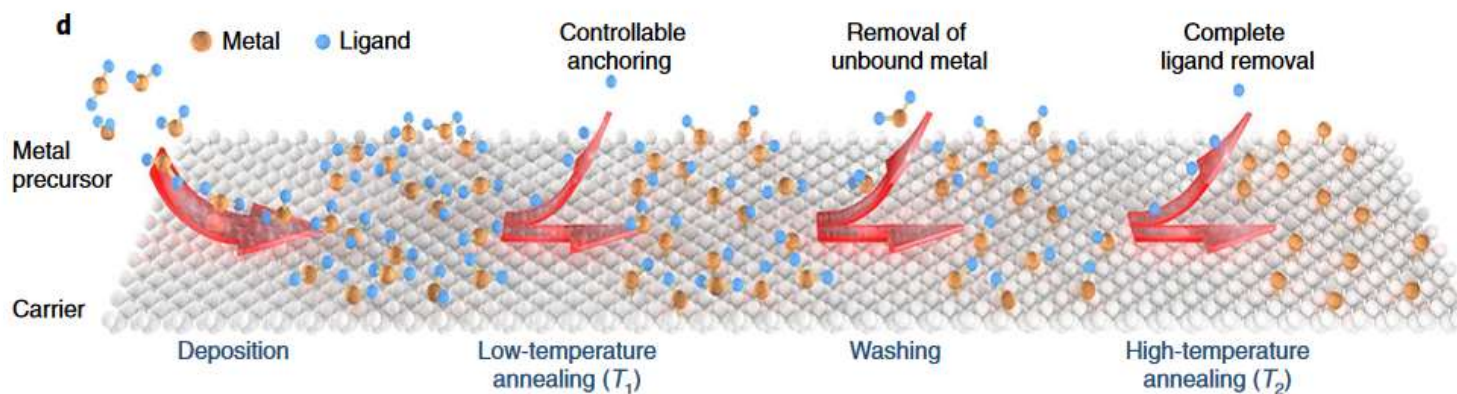
nature
nanotechnology



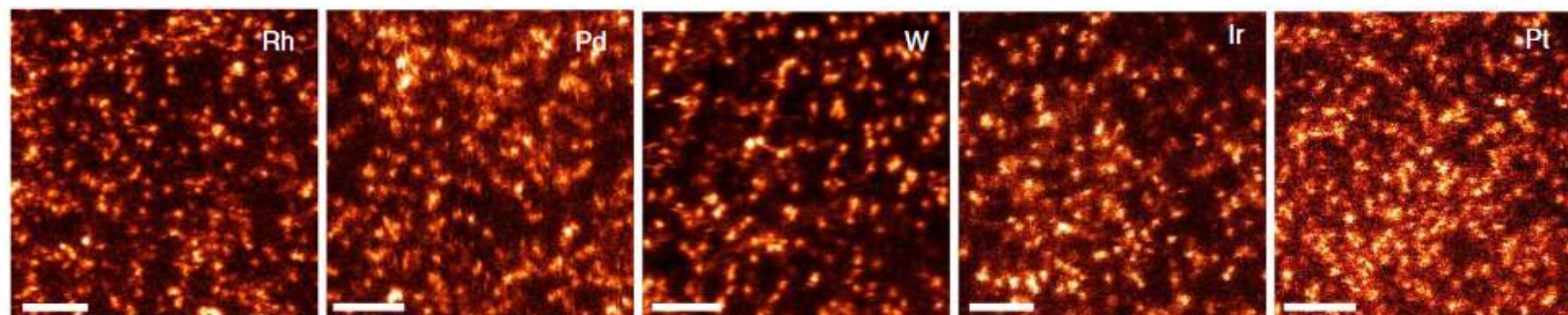
Scalable two-step annealing method for preparing ultra-high-density single-atom catalyst libraries

Xiao Hai ^{1,11}, Shibo Xi ^{2,11}, Sharon Mitchell ^{3,11}, Karim Harrath^{4,5,11}, Haomin Xu ¹, Dario Faust Akl³,
Debin Kong⁶, Jing Li ¹, Zejun Li¹, Tao Sun¹, Huimin Yang ¹, Yige Cui¹, Chenliang Su ⁷, Xiaoxu Zhao ⁸✉,
Jun Li ^{4,5}✉, Javier Pérez-Ramírez ³✉ and Jiong Lu ^{1,9,10}✉

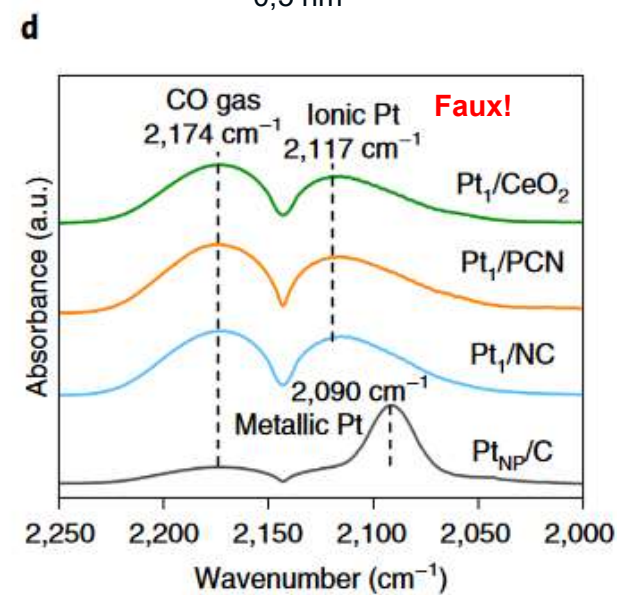
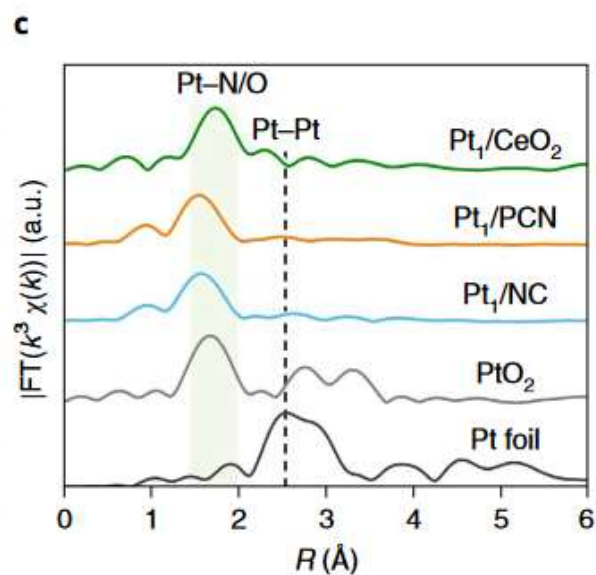
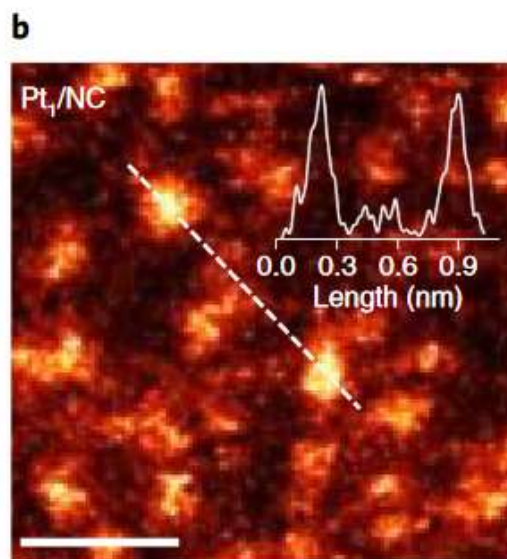
NATURE NANOTECHNOLOGY | VOL 17 | FEBRUARY 2022 | 174–181 | www.nature.com/naturenanotechnology



NC: carbone dopé à N; PCN: nitrure de carbone polymérique



0,5 nm





Economie de la synthèse

Quelques dizaines de pages plus haut...

SINGLE-ATOM CATALYSIS

All the lonely atoms, where do they all belong?

A spectacular advance in catalyst synthesis is demonstrated by a two-step process that yields high loadings of single atoms at a kilogram scale suitable for industrial deployment.

John R. Regalbuto and Abhaya K. Datye

NATURE NANOTECHNOLOGY | VOL 17 | FEBRUARY 2022 | 109-111 | www.nature.com/naturenanotechnology



Economie de la synthèse

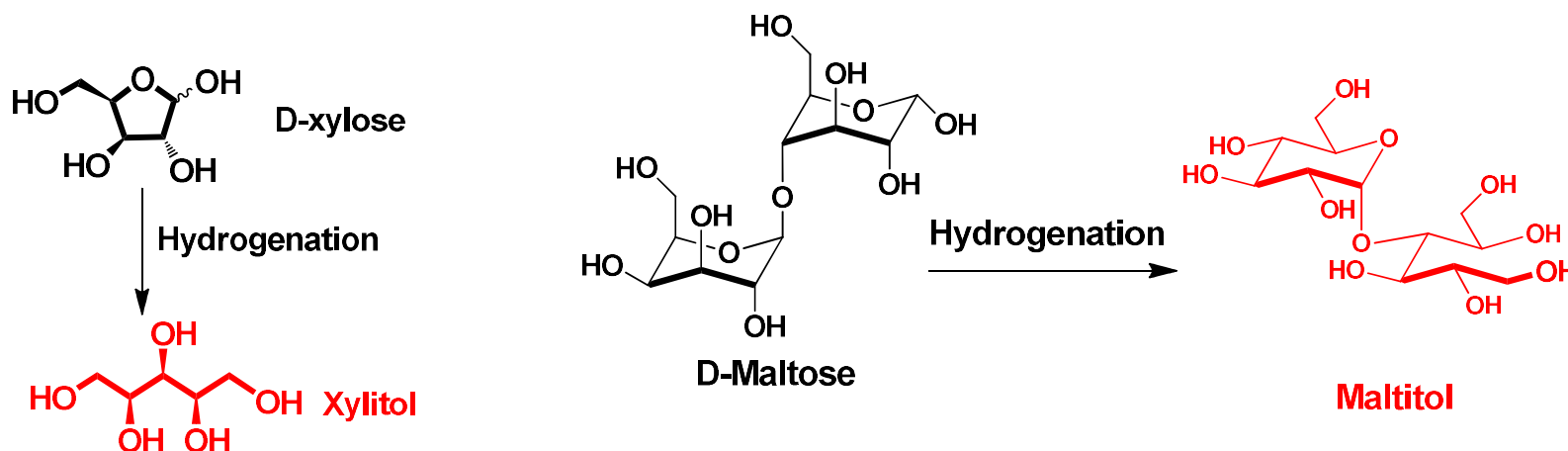
More work is needed to validate some of the critical steps in Hai et al.'s approach. For example, improved CO infrared spectroscopy characterization, in which the CO gas-phase background is subtracted, could more clearly reveal the adsorbed species, as has been shown with single-atom platinum catalysts of much lower weight loading^{4,9}. The high metal loading achieved here should make this characterization much easier. Post-reaction characterization of additional metals will further confirm the general stability of atoms synthesized by this method. While the authors demonstrate scale up to kilogram-sized batches, the washing

step might create a waste stream of dilute catalyst precursors, which should be minimized or eliminated.

The ability to stabilize high densities of single atoms is a holy grail of catalyst synthesis, from which point many other controllable syntheses become possible. A great palette of fundamental study of atom engineering now awaits: can the density of atoms and the degree of interaction between single sites or multiple metals be controlled by the degree of nitrogen doping in a carbon surface or oxygen defect formation? Are there better anchoring sites? Can the distance between anchored atoms be tuned to change reactivity, and how might this be done? What about engineering interwoven, dense patterns of different metals to achieve bifunctionality? What are the limits of stability? On the practical side, the commercial potential for single-atom catalysts has already been recognized¹⁰. Higher densities of catalyst sites make their potential all the more relevant.

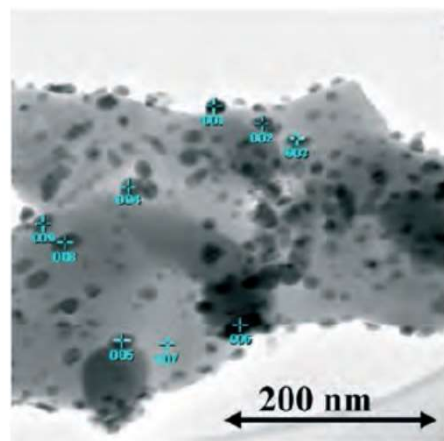
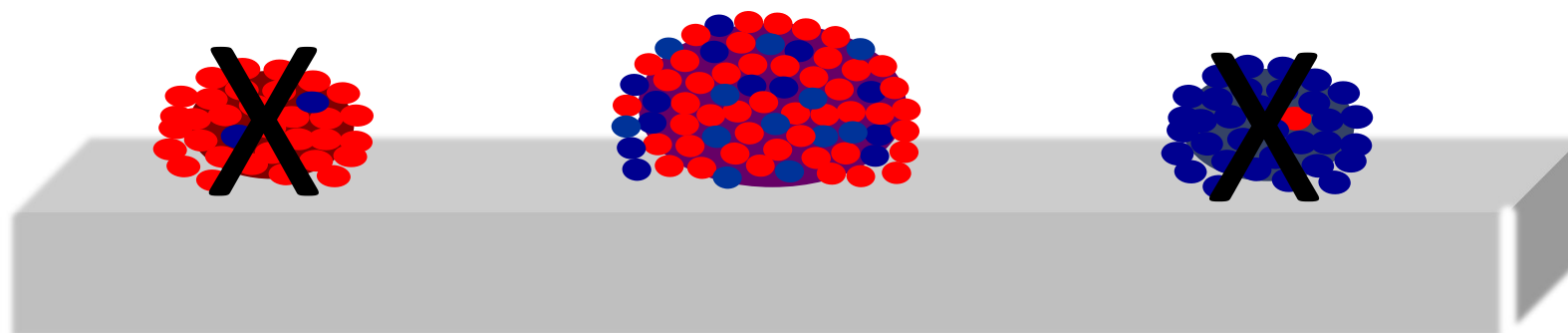
Du côté des métaux non nobles :

- Catalyse d'hydroconversion sélective de molécules biosourcées



- Remplacement des métaux nobles ou du nickel de Raney par des combinaisons **Ni-Fe** à base de deux métaux abondants

➤ Verrou: **contrôle de la taille, de la composition ET de l'ordre chimique** des nanoparticules Ni-Fe



EDX analysis

Pos.	Fe/Ni
001	48/52
002	14/86
003	6/94
004	17/83
005	68/32
006	61/39
007	0/0
008	68/32
009	72/28

Co-imprégnation de $\text{Ni}(\text{NO}_3)_2$ et $\text{Fe}(\text{NO}_3)_3$ sur $\alpha\text{-Al}_2\text{O}_3$

Wang et al., Appl. Catal. A 392 (2011) 248



← Ar

← SiO₂

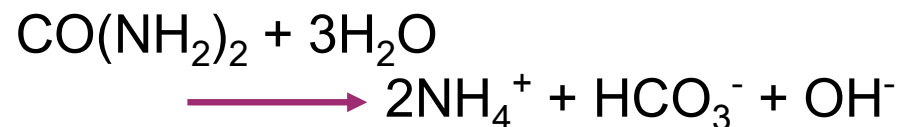
← H₂SO₄

← NiSO₄•6H₂O

← CO(NH₂)₂

← FeSO₄•7H₂O

Dépôt-précipitation à l'urée



Formation d'un phyllosilicate mixte
réduit par H₂ en Ni-Fe/SiO₂



Ar

SiO₂

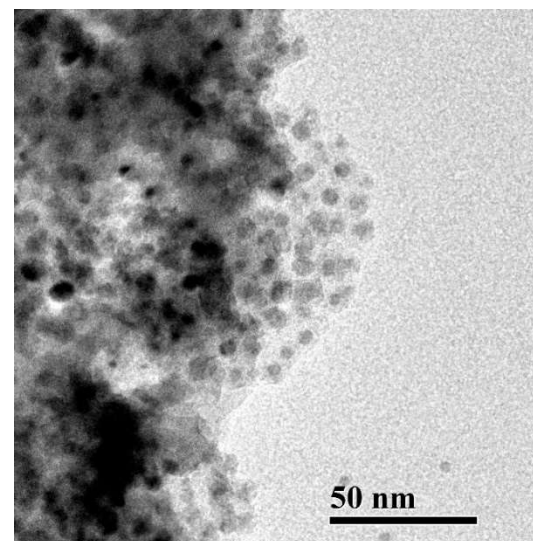
H₂SO₄

NiSO₄·6H₂O

CO(NH₂)₂

FeSO₄·7H₂O

Fonctionne...



Ni-Fe/SiO₂ après réduction

Shi et al., Catal. Today 334 (2019) 162



← Ar

← SiO₂

← H₂SO₄

← NiSO₄•6H₂O

← CO(NH₂)₂

← FeSO₄•7H₂O

Fonctionne...

mais

**30% du Ni
et 60% du Fe
restent en solution
en fin de synthèse !**

Shi et al., Catal. Today 334 (2019) 162



Economie de la synthèse

Une synthèse de matériau catalytique doit-elle être jugée comme pertinente en fonction de son **rendement en métal utile**

et de la **stabilité de ce rendement après utilisation**

... et ce d'autant plus que le métal est coûteux?



Economie de la synthèse

Une synthèse de matériau catalytique doit-elle être pensée avec sa **stratégie de recyclage ou de régénération**?

Synthesis and Regeneration of Ni-Phyllosilicate Catalysts Using a Versatile Double-Accelerator Method: A Comprehensive Study

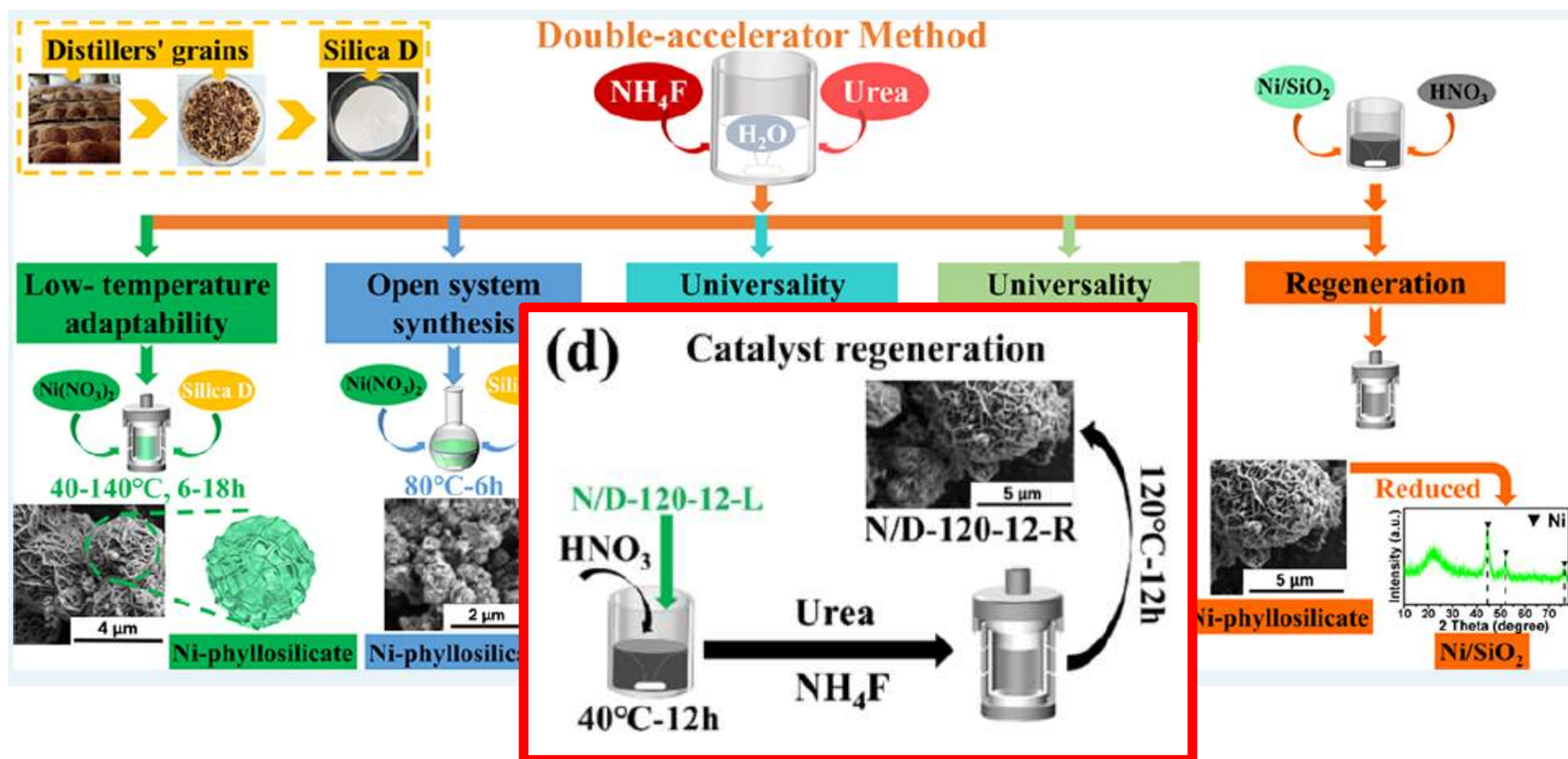
Yaqi Chen and Qing Liu*



Cite This: *ACS Catal.* 2021, 11, 12570–12584



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Verrou 2

Analyses locales:

Echelle atomique
Hétérogénéité



Analyses locales

More work is needed to validate some of the critical steps in Hai et al.'s approach. For example, improved CO infrared spectroscopy characterization, in which the CO gas-phase background is subtracted, could more clearly reveal the adsorbed species, as has been shown with single-atom platinum catalysts of much lower weight loading^{4,9}. The high metal loading achieved here should make this characterization much easier. Post-reaction characterization of additional metals will further confirm the general stability of atoms synthesized by this method. While the authors demonstrate scale up to kilogram-sized batches, the washing

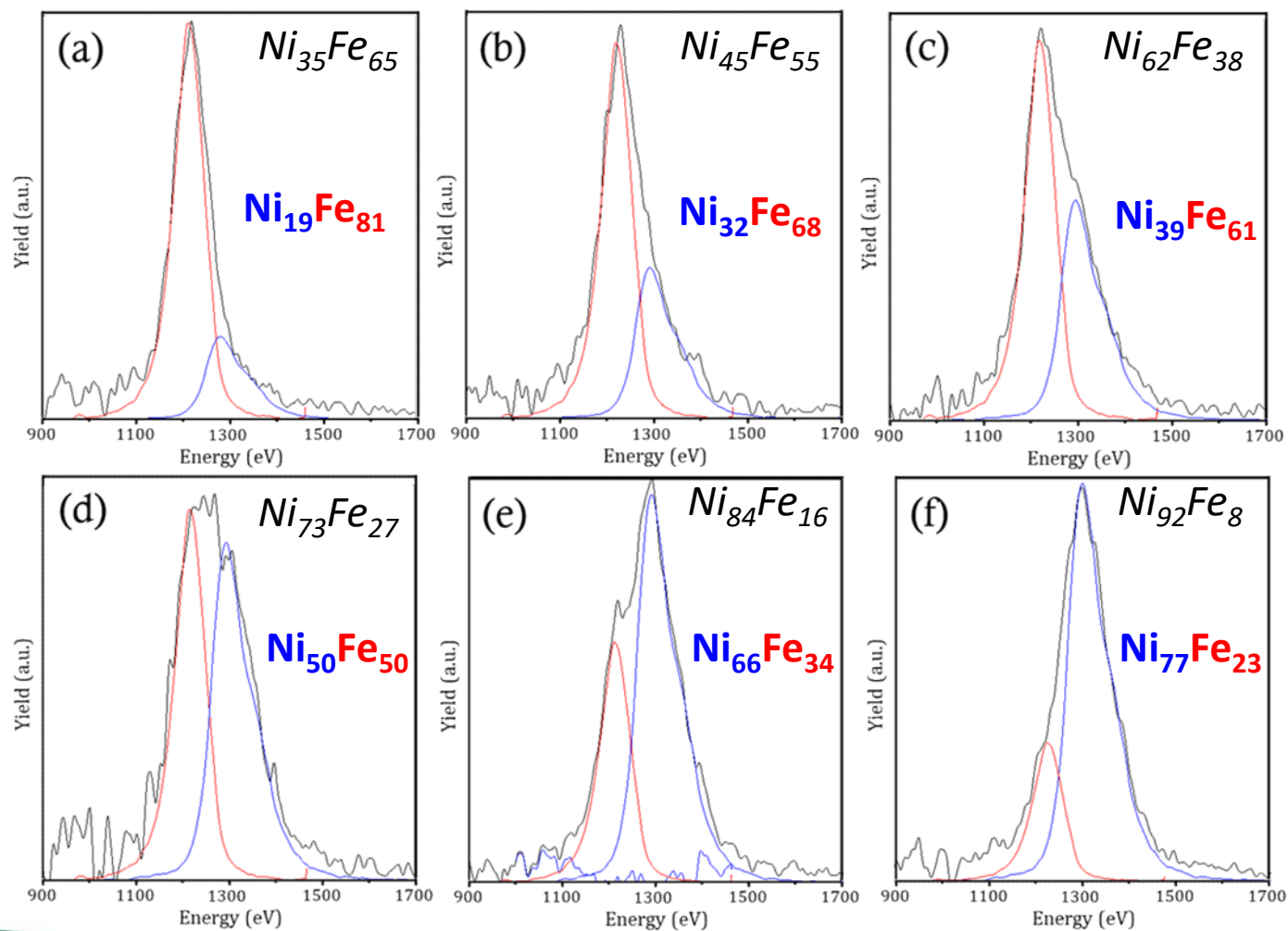
step might create a waste stream of dilute catalyst precursors, which should be minimized or eliminated.

The ability to stabilize high densities of single atoms is a holy grail of catalyst synthesis, from which point many other controllable syntheses become possible. A great palette of fundamental study of atom engineering now awaits: can the density of atoms and the degree of interaction between single sites or multiple metals be controlled by the degree of nitrogen doping in a carbon surface or oxygen defect formation? Are there better anchoring sites? Can the distance between anchored atoms be tuned to change reactivity, and how might this be done? What about engineering interwoven, dense patterns of different metals to achieve bifunctionality? What are the limits of stability? On the practical side, the commercial potential for single-atom catalysts has already been recognized¹⁰. Higher densities of catalyst sites make their potential all the more relevant.

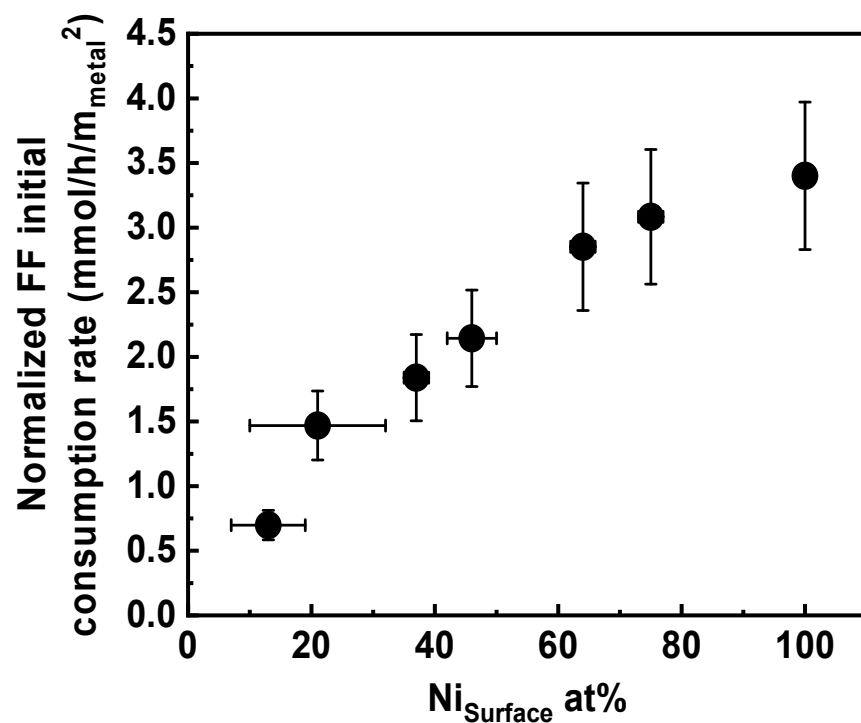


Analyses locales

Analyse de surface de Ni-Fe/SiO₂ par spectroscopie LEIS

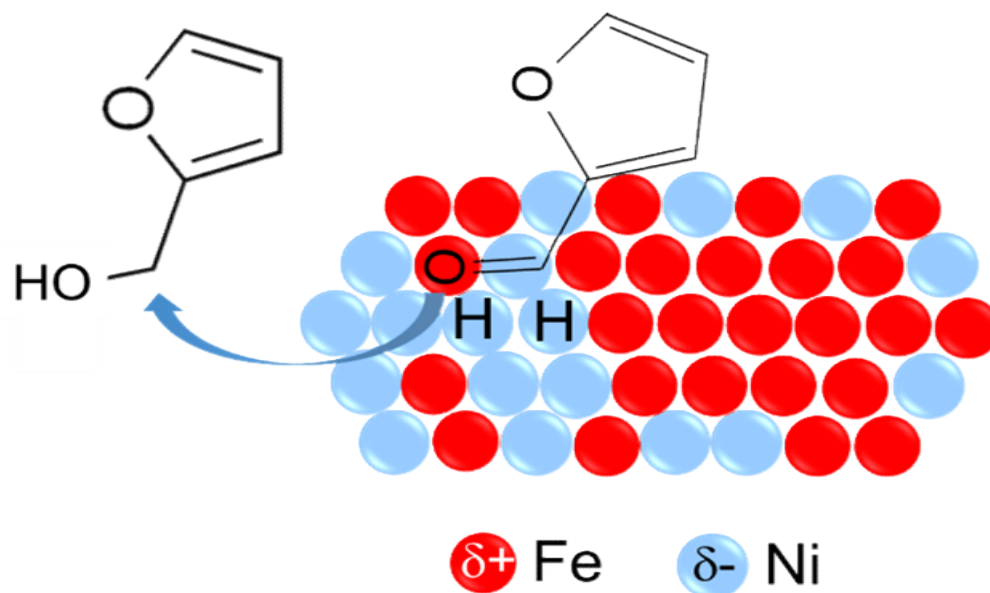


Catalyse: une répartition sans doute hétérogène de Ni



Vitesse initiale de consommation du furfural par hydroconversion à 150 ° C dans l'isopropanol = f (Ni surface at %)

Des îlots de Ni dilués par Fe?



Shi et al., Chem. Catal. 2 (2022), <https://doi.org/10.1016/j.checat.2022.04.009>

Shi et al., L'Actualité Chimique, 473-474 (2022) 62



Analyses locales



Commentaires de Jean-Luc Rousset (IRCELYON)

Entropie favorable au mélange Ni-Fe

Energie de surface favorable à Ni

?

Quelles perspectives pour la **modélisation de systèmes à plusieurs centaines d'atomes** en interaction partielle avec un **support**, sous **pression de réactifs**?



Analyses locales

Pour la compréhension des performances du matériau catalytique, sa synthèse peut-elle se concevoir
autour d'un élément de mesure interne?

Operando Nanoscale Sensors in Catalysis: All Eyes on Catalyst Particles

Thomas Hartman, Robin G. Geitenbeek, Caterina S. Wondergem, Ward van der Stam, and Bert M. Weckhuysen*

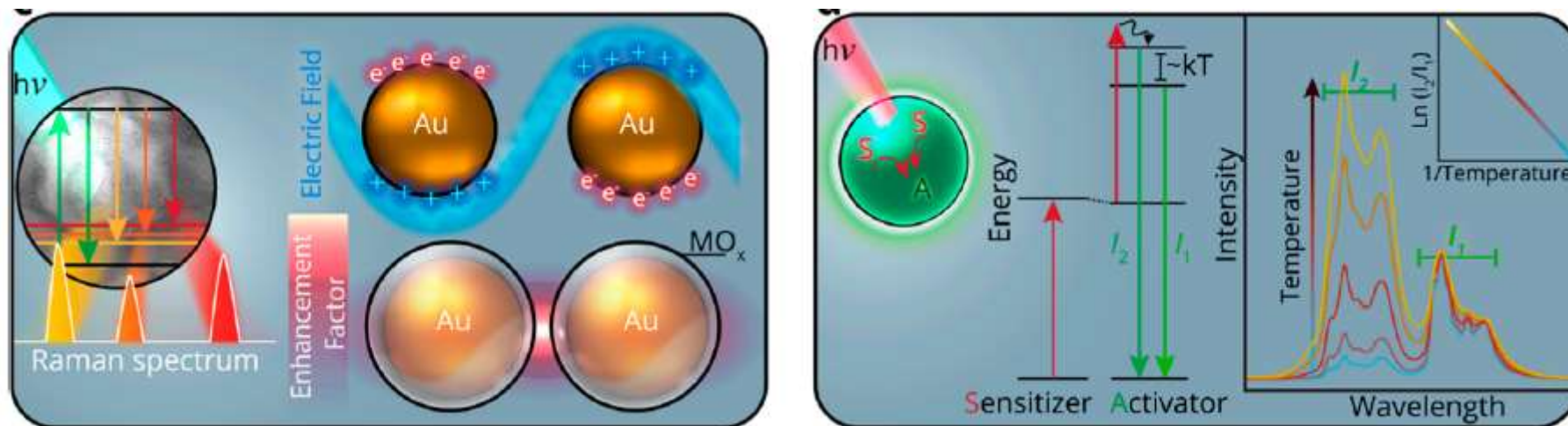


Cite This: *ACS Nano* 2020, 14, 3725–3735



Read Online

ABSTRACT: An era of circularity requires robust and flexible catalysts and reactors. We need profound knowledge of catalytic surface reactions on the local scale (*i.e.*, angstrom–nanometer), whereas the reaction conditions, such as reaction temperature and pressure, are set and controlled on the macroscale (*i.e.*, millimeter–meter). Nanosensors operating on all relevant length scales can supply this information in real time during *operando* working conditions. In this Perspective, we



Schematic illustration of the working principles of plasmon-enhanced Raman spectroscopy (PERS). Raman spectroscopy probes the inelastic scattering of light after excitation of matter using monochromatic light to obtain molecular structures. The Raman scattering intensity and the incident light can be enhanced by plasmonic nanoparticles by exciting the dipolar localized surface plasmon resonance. This Raman enhancement can be strongly improved in hot spots between two or more shell-isolated nanoparticles. (d) Schematic illustration of the working principles of luminescence thermometry. Absorption of near-infrared light results in the emission of upconverted green light using lanthanide sensitizers and activators. Temperature-dependent luminescence is governed by Boltzmann statistics; the logarithm of the luminescence intensity ratio plotted *versus* $1/T$ yields a linear correlation; the steepness of the fitted line corresponds to the energy difference between the excited states.

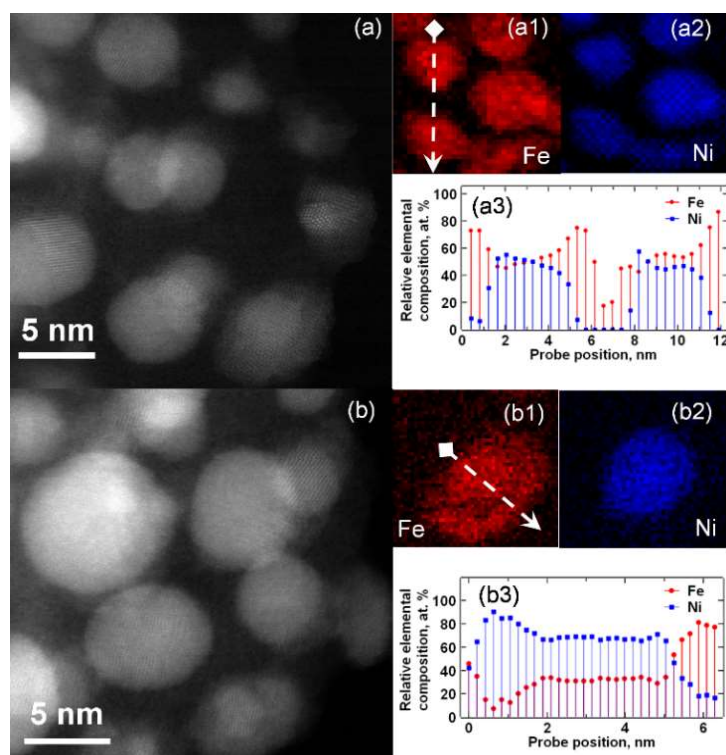


Verrou 3

Représentativité:

Echantillonnage
Statistiques
Analyse des données

Mise en évidence par STEM-HREELS de l'enrichissement en Fe en bord de nanoparticules



≈ 10 particules par catalyseur





Représentativité

Commentaire d'un reviewer:

Etape 1: “ In Table S4, significant variability exists in the composition of the nanoparticles measured... **How confident are the authors** that the particles are **sufficiently uniform** to make average catalytic correlations? “

Etape 2: “ The authors... do not provide any additional characterization beyond citing their previous work. **I think it is essential to obtain more STEM-EDS (sic) measurements of individual nanoparticles for the three new samples studied in this paper “**

?



Représentativité

Beaucoup d'analyses sur différentes particules...

Mimicking industrial aging in fluid catalytic cracking: A correlative microscopy approach to unravel inter-particle heterogeneities



M. Gambino^a, A.E. Nieuwelink^a, F. Reints^a, M. Veselý^a, M. Filez^a, D. Ferreira Sanchez^b, D. Grolimund^b, N. Nesterenko^c, D. Minoux^c, F. Meirer^{a,*}, B.M. Weckhuysen^{a,*}

^a Inorganic Chemistry and Catalysis, Department of Chemistry, Utrecht University, Universiteitsweg 99, 3584 CG Utrecht, the Netherlands

^b Swiss Light Source, Paul Scherrer Institute, Villigen, Switzerland

^c Total Research and Technology Feluy, Seneffe, Belgium

Journal of Catalysis 404 (2021) 634–646

Représentativité

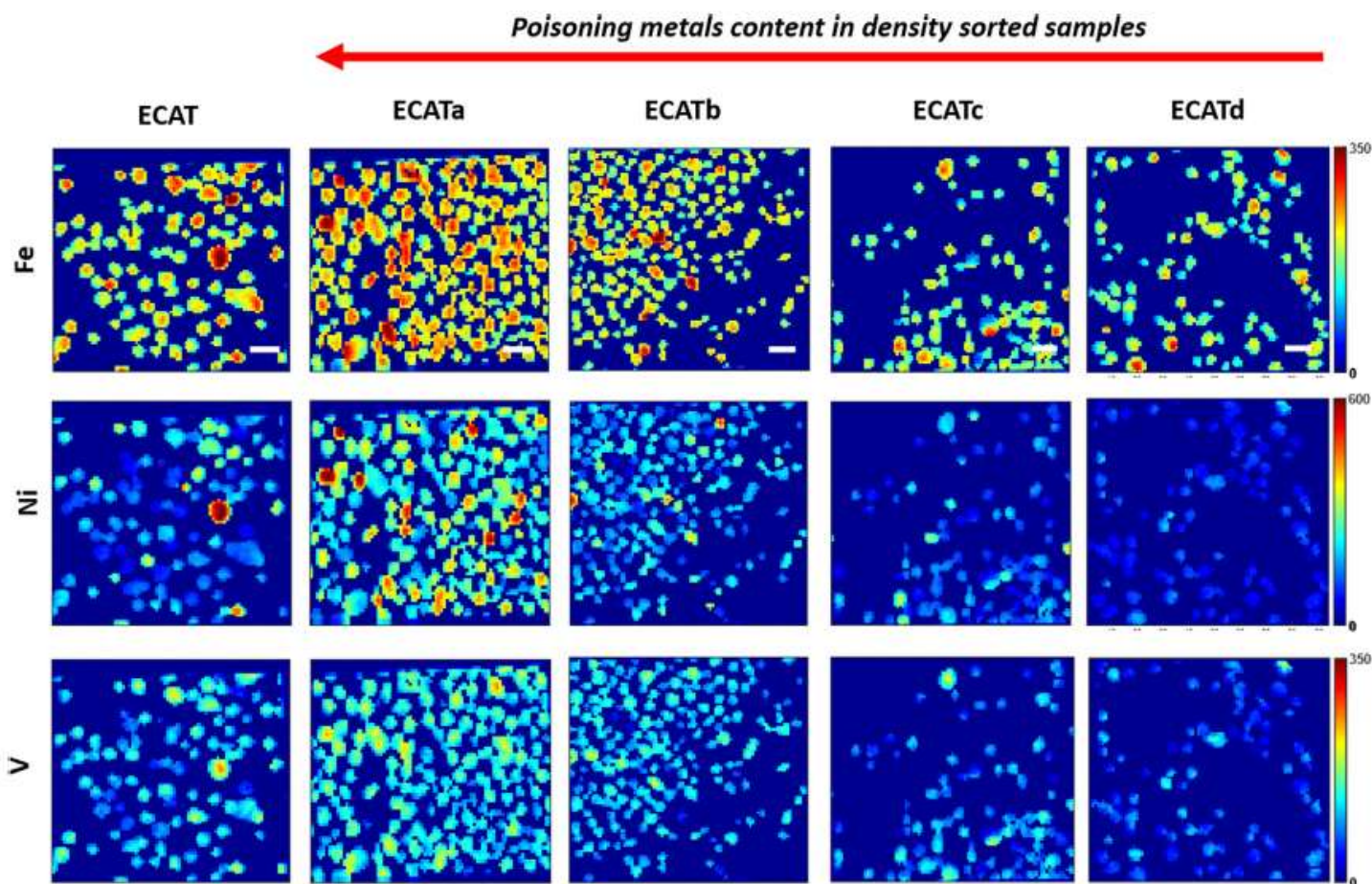


Fig. 3. Elemental distribution maps of Fe, Ni and V poisoning metals for the different ECATs samples, obtained after batch fitting analysis of the μ XRF dataset. The scale bar indicates 150 μ m.

... mesures à l'échelle de la **centaine de microns**



Représentativité

... ou une analyse très poussée d'une seule particule?

3-D X-ray Nanotomography Reveals Different Carbon Deposition Mechanisms in a Single Catalyst Particle

Martin Veselý,^[a] Roozbeh Valadian,^[a] Leon Merten Lohse,^[b] Mareike Toepperwien,^[b] Kathryn Spiers,^[c] Jan Garrevoet,^[c] Eelco T. C. Vogt,^[a, d] Tim Salditt,^[b] Bert M. Weckhuysen,^{*,[a]} and Florian Meirer^{*,[a]}

ChemCatChem 2021, 13, 2494–2507





Représentativité

Comment définir l'**équilibre entre le “coût appareil” (€, t) et une description jugée représentative du matériau** à l'échelle souhaitée?

Le futur chercheur / la future chercheuse sur les matériaux catalytiques doit-il/elle être également formé/formée **en codage et en traitement de données?**



Représentativité

Laboratoire interdisciplinaire



Formation interdisciplinaire
(master universitaire / écoles)



Pérennisation des compétences



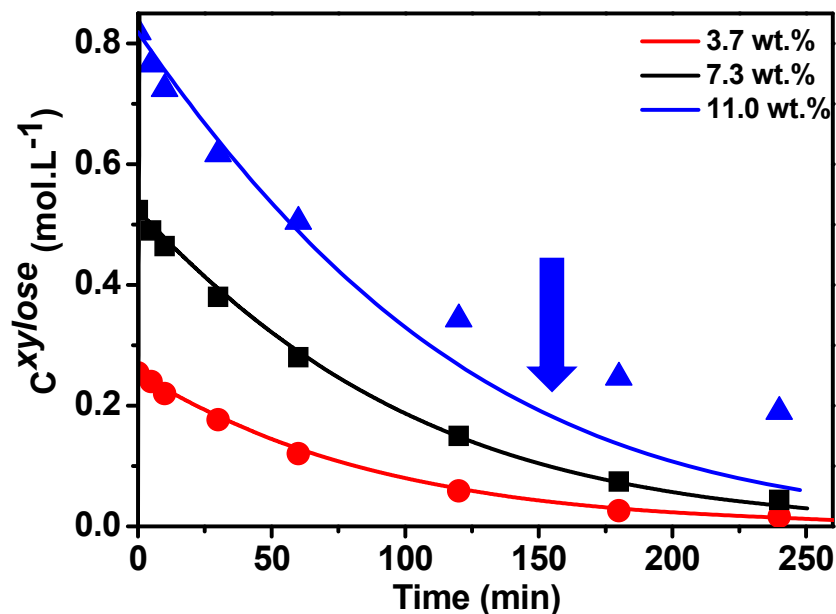
Verrou 4

Stabilité:

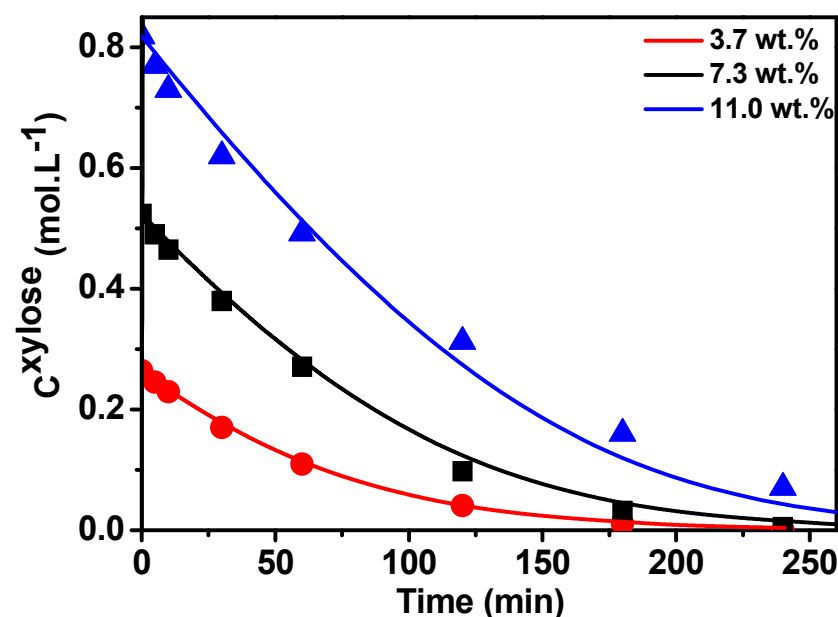
Phase liquide
Analyse in situ

Performances en hydrogénation du xylose en xylitol dans l'eau

Ni₁₀₀/SiO₂

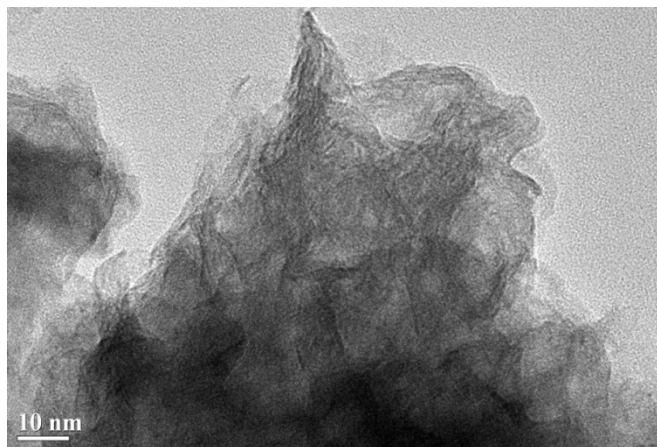


Ni₆₂-Fe₃₈/SiO₂

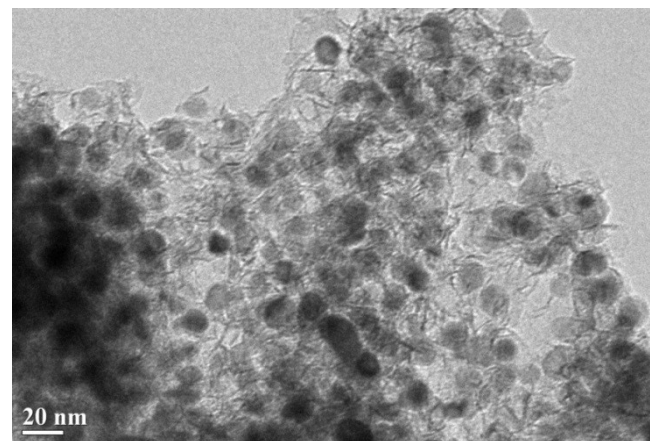


symboles: mesures expérimentales; courbes: modèle cinétique

Sadier et al., Appl. Catal. B, 298 (2021) 120564

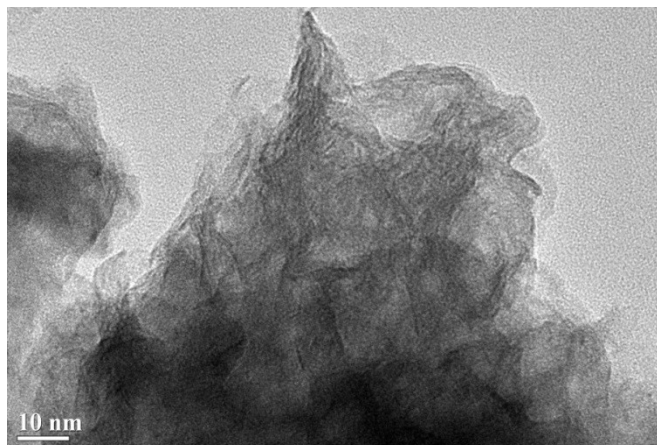


Retour à l'état phyllosilicate



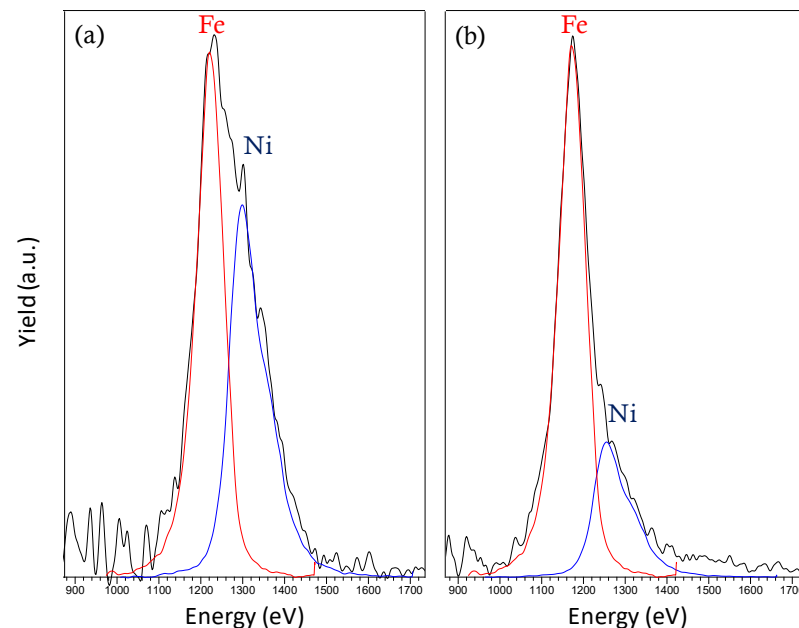
Stable en apparence...

$\text{Ni}_{100}/\text{SiO}_2$



Retour à l'état phyllosilicate

$\text{Ni}_{62}\text{-Fe}_{38}/\text{SiO}_2$



LEIS, avant....

...LEIS, après

...mais enrichissement en Fe en surface

Le **suivi de l'évolution du matériau en conditions de réaction est indispensable** pour tirer des conclusions pertinentes...

... mais comment suivre la **cinétique d'évolution** d'un metal supporté :

au sein d'un système triphasique

dans lequel le solide est sous forme granulaire ($\neq$ électrochimie)

sous pression

dans un réacteur à forte inertie thermique?



Ingénierie

In Situ Microfocus Chemical Computed Tomography of the Composition of a Single Catalyst Particle During Hydrogenation of Nitrobenzene in the Liquid Phase**

Stephen W. T. Price,* Kalotina Geraki, Konstantin Ignatyev, Peter T. Witte, Andrew M. Beale,* and J. Fred W. Mosselmans

Angew. Chem. 2015, 127, 10024–10027

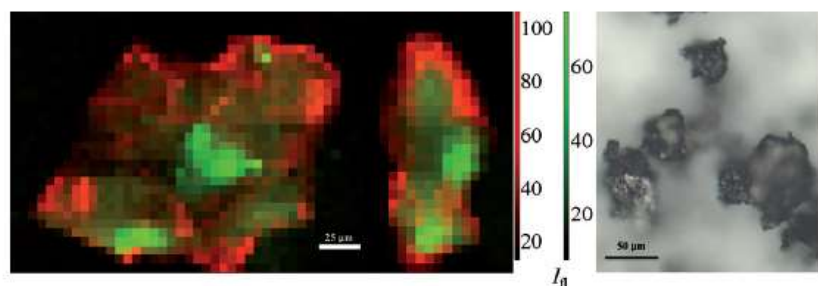


Figure 1. Left: Orthogonal XRF maps (at 20.3 keV) of a particle under operating conditions (ethanol, 22.3 mM nitrobenzene, and 1 bar H₂) after collection of μ-XRF-CT and μ-XRD-CT sinograms. Elemental distributions throughout the carbon support are shown in red (Pt) and green (Mo). Each pixel is 5 × 5 μm. Red and green scale bars at the side show fluorescent signal intensity (I_{fl}), white scale bar (in the image) = 25 μm. Right: Photo of typical catalyst particles (ex situ); scale bar = 50 μm.

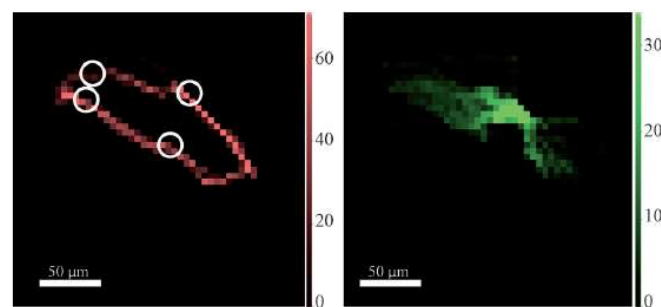
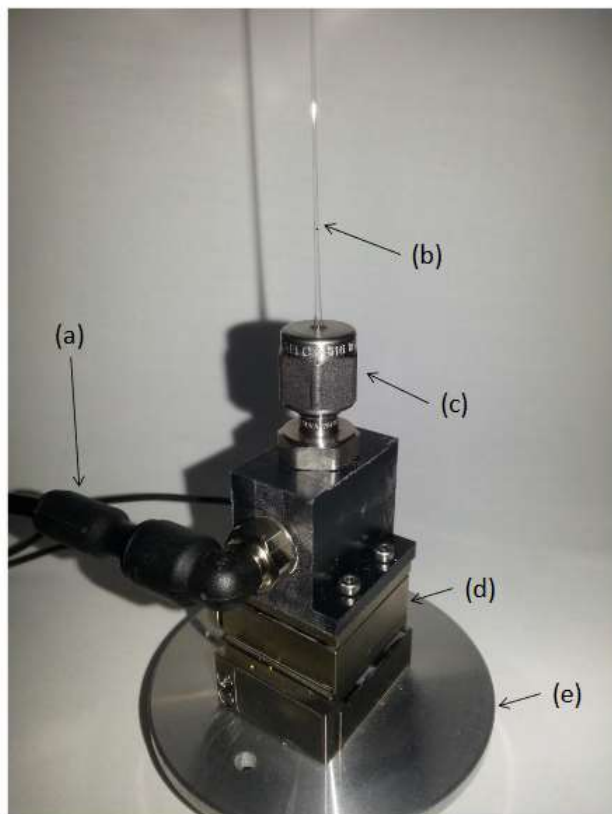


Figure 2. Cross-sections from μ-XRF-CT at 20.3 keV showing Pt (red, left) and Mo (green, right) distributions throughout the carbon support. Each pixel is 5 × 5 μm. Data collected under operating conditions (ethanol, 22.3 mM nitrobenzene, and 1 bar H₂). The slice corresponds to a horizontal cross-section 60 μm from the bottom of the particle (see Figure 1). The white circles correspond to the approximate locations of the Pt crystals (identified from the 111 reflection). Scale bars at the sides show fluorescent signal intensity (I_{fl}), white scale bars (within the images) = 50 μm.



SI Figure 11: Microreactor for in situ tomography. (a) gas inlet, (b) single catalyst particle mounted in 0.4 mm diameter glass capillary, (c) capillary sealed with two o-rings, (d) nano-positioning motors for centring capillary, (e) base to connect to beamline rotation stage.

A single particle was mounted in 0.4 mm OD borosilicate glass capillary (0.01 mm wall thickness) using quartz wool and H_2 purged ethanol was injected into the capillary (circa 10 μL) then pressurized to 1 bar with H_2 .

However, the activity was moderated (23°C and 1 bar H_2 , with no stirring because of the capillary dimensions) to suit the conditions for performing tomography so as to allow for real-time imaging during steady-state catalytic turnover. In all particles chosen in this and in our previous study,^[1b] the elemental distribution is consistent, and does not appear to be influenced by particle size. It should be noted that with the lack of stirring, diffusion becomes the main mass-transport process of reactant/product and therefore may influence the activity of the catalyst, and also the ratio of products detected by Raman spectroscopy.

Operando X-ray Absorption Spectroscopy Studies of Sintering for Supported Copper Catalysts during Liquid-phase Reaction

Brandon J. O'Neill,^[a] Jeffrey T. Miller,^[b] Paul J. Dietrich,^[c] Fred G. Sollberger,^[c] Fabio H. Ribeiro,^[c] and James A. Dumesic^{*[a]}

ChemCatChem **2014**, *6*, 2493 – 2496

For *operando* reactions, catalyst samples were diluted by half with silica to improve X-ray transmission and loaded into a glassy C tube reactor (10 mm x 4 mm x 200 mm OD x ID x L).^[2] Next, the catalysts were reduced at 573 K at the beamline, cooled to room temperature, pressurized to 22 bar with He at 30 cm³ (STP)/min and checked for leaks. Once the reactor was determined to be leak free, the reactor was depressurized, re-pressurized to 22 bar with H₂ at 30 cm³ (STP)/min, and then heated to the reaction temperature of 403 K. Once the reaction temperature was reached, the flow of the liquid reaction feed (5 wt% furfural in 1-butanol) was started using a syringe pump (Teledyne-Isco). The X-ray beam was 0.5 x 0.3 mm², and data were collected in quick scan transmission mode. Each spectrum was obtained in about 30 sec and approximately 10 spectra were obtained at each reaction time. Gas bubbles from the H₂ co-feed in the liquid created irregularities in the spectrum which could not be removed during data processing. It was important to minimize these effects, so the gas flow was reduced (0-5 cm³ (STP)/min) during XAS data acquisition. Only spectra free of these features were averaged,

Continuous-flow reactor setup for *operando* x-ray absorption spectroscopy of high pressure heterogeneous liquid-solid catalytic processes

Cite as: Rev. Sci. Instrum. 92, 124101 (2021); doi: [10.1063/5.0057011](https://doi.org/10.1063/5.0057011)

Submitted: 17 May 2021 • Accepted: 13 November 2021 •

Published Online: 2 December 2021



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Benedikt J. Deschner,^{1,a)}  Dmitry E. Doronkin,^{2,3,a)}  Thomas L. Sheppard,^{2,3}  Georg Rabsch,¹
Jan-Dierk Grunwaldt,^{2,3}  and Roland Dittmeyer^{1,2} 

saturation préalable du liquide avec le gaz

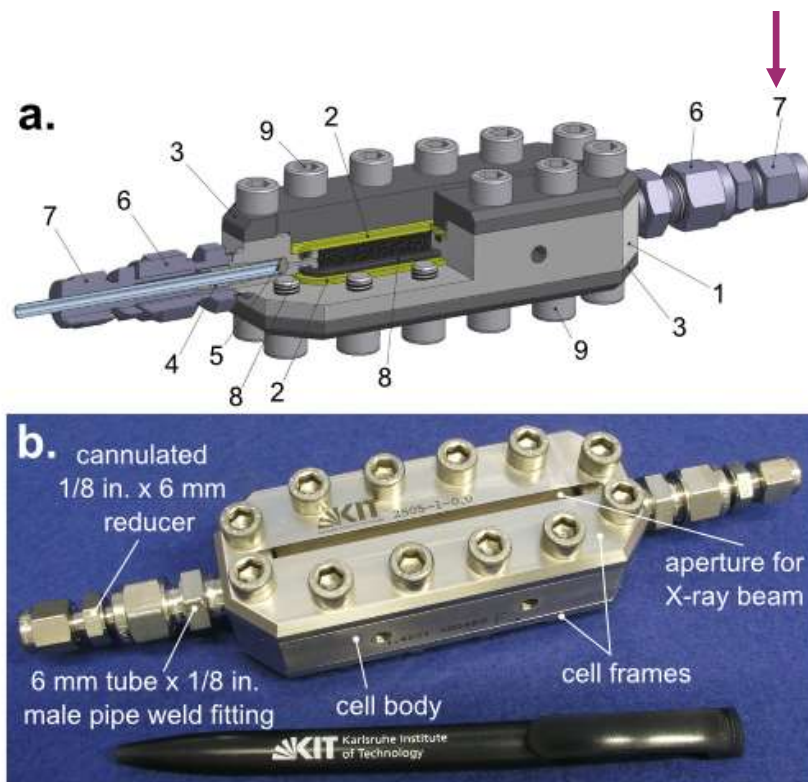


FIG. 1. (a) Drawing of the assembled *in situ* continuous-flow XAS cell. The numbers refer to list parts as described in Table S2. (b) Photograph of the manufactured cell.

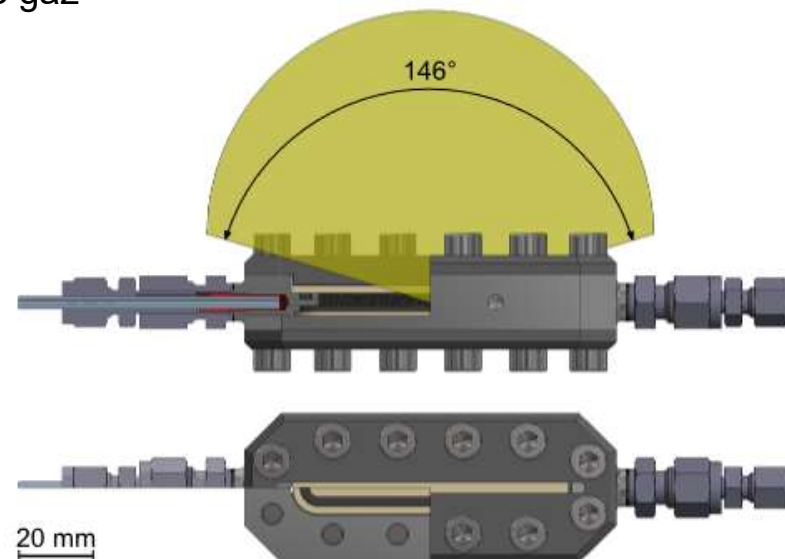


FIG. 3. Accessibility of the outgoing x-ray beam over a wide angular range, allowing transmission, fluorescence, and diffraction studies on the catalyst.

Provided the catalyst is supported on a light-atom material, our cell equipped with 2 mm PEEK windows allows XAS studies at the Pt L₃ (11.56 keV) and Au L₃ (11.92 keV) edges and—with thinner 1 mm windows—reaches as low as the Cu K edge (8.98 keV) (see Fig. 6).



Conclusions: verrous

Economie de la synthèse

Matériau catalytique

Analyse locale des hétérogénéités

Représentativité de l'analyse

Stabilité en milieu bi/triphasique "réaliste"



Conclusions: descripteurs, outils, pistes

Rendement en metal utile

Matériau catalytique

Modélisation / Éléments de mesure internes

Coût (€, t) / Formation

Ingénierie



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Thank you for your attention!